COLLEGE OF SCIENCE CHFMISTRY

Synthetic and Biological Approaches for Tuning Reactivity of First-Row Transition Metals

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We seek to combine inorganic, organometallic, and bioinorganic tools to enable advancements in energy- and biologically-relevant problems. We aim to leverage metal-metal and metal-ligand cooperativity using synthetic and biological scaffolds to manipulate the electronic structure and reactivity of earth-abundant, first-row transition metals and promote challenging transformations. Spectroscopic, mechanistic, and reactivity studies will help inform the development of efficient catalysts.

titanium 22	vanadium 23	chromium 24	manganese 25	iron 26	cobalt 27	nickel 28	copper 29
Ti	V	Cr	Mn	Fe	Со	Ni	Cu
47.867	50.942	51.996	54.938	55.845	58.933	58.6934	63.546

Frustrated transition metal pairs Alkane dehydrogenation



Electronic structure / function correlation Aerobic alkane hydroxylation



Proteins as ligands for unique metallocofactors Small molecule activation and catalysis





Frustrated Transition Metal Pairs

Frustrated transition metal pairs (FTMPs) for C–H activation

Right. We seek to explore bimetallic cooperativity to induce polarization and promote activation of strong bonds. We are interested in designing and exploring first-row transition metal systems that expand upon the concept of Frustrated Lewis Pairs (FLPs).





Below. We aim to systematically tune these FTMPs to elaborate the extent of polarization and its impact on the reactivity of each metal site. Ultimately, we seek to provide an alternate strategy for activation of strong aliphatic C–H bonds using earth-abundant first-row transition metals.



What is the role of M–M

Bioinspired Dinuclear Complexes for Aerobic Hydroxylation

Enzymatic and industrial aerobic methane oxidation – Fe, and Cu, sites



Left. Direct aerobic oxidation of methane to methanol remains a significant target in the synthetic community. Systems such as methane monooxygenase enzymes (e.g., sMMO) and copper zeolites (e.g., Cu-ZSM-5) demonstrate such exquisite Studies have reactivity. proposed bimetallic bridging oxo units as the active species responsible for the challenging C-H cleavage, albeit an understanding of their intrinsic reactivity is still lacking.

What is the correlation between electronic structure and reactivity of dimeric oxo species?

Bis-alkoxide bimetallic platforms to support reactive bimetallic oxo species

interactions?

What is the reactivity profile of FTMPs?

Can we tune the FTMPs to accomplish C–H activation?

Above. We will be targeting both pairs of monometallic complexes as well as design dinuclear platforms to (1) enable synthesis of heterobimetallic complexes; (2) control the M–M separation; and (3) accommodate substrate interactions with the metal sites.





What are the salient electronic features for reactive $M_2(\mu-O)$ and $M_2(\mu-O)_2$ units?

What is the influence of spin state, oxidation state, and geometry?

Proteins as Ligands for Unique Metallocofactors

*Reactivity of a His*₅ *metal binding site*





Right. We want to leverage proteins to model organometallic complexes and take advantage of secondary sphere interactions within the protein structure to



Group transfer and dehydrogenation reactivity

